Journal of Advanced Biomedical and Pharmaceutical Sciences

Journal Homepage: http://jabps.journals.ekb.eg



Comparative Head Space GC/MS Studies of Different Flavored Moâssel in the Egyptian Market (II)

Marwa M. Ismail¹, Ashraf N. E. Hamed^{1*}, Mostafa A. Fouad¹, Mohamed S. Kamel^{1,2}

¹ Department of Pharmacognosy, Faculty of Pharmacy, Minia University, 61519 Minia, Egypt

² Department of Pharmacognosy, Faculty of Pharmacy, Deraya University, 61111 New Minia, Egypt

Received: March 9, 2019; revised: April 11, 2019; accepted: April 23, 2019

Abstract

This study designed to investigate three various samples of Egyptian Moâssel (flavored tobacco) used in water pipe smoking. The specimens (Guava, Mixed Fruits and Watermelon) were collected from the Egyptian market, which were produced by Al Dandash company (Egyptian famed tobacco company). They were examined by Head Space GC/MS. There were relatively differences among the investigated specimens. The identified compounds of the first one (Guava) exhibited 31 ingredients, which represented (81.38 %) of the compounds. The main one was fraistone (12.87 %). While, the second one (Mixed Fruits) specimen displayed 36 recognized compounds, which represented (91.98 %) of the sample components. The main constituent was isoamyl acetate (19.83 %). Finally, the last one (Watermelon) exhibited 27 identified ingredients, which represented (68.60 %) of the total compounds and the highest compound was 1,2-propanediol (25.23 %) of the constituents.

Key words

Egyptian Flavored Moâssel, Head Space GC/MS, Al Dandash Company

1. Introduction

Smoking of Tobacco is a serious and addictive habit. In the UK, half of all lifelong cigarette smokers died due to smoking [1]. Moreover, smokers reduced an average of around 3 months of life expectation for every year smoked after the age of 35; in sustained smokers this amounts to a total loss of around 10 years of lifetime [1, 2]. Furthermore, passive inhalation of the smoke harms others [3, 4]. One of tobacco smoking types is Shisha. It has many alternative names such as Hookah or water pipe or hubble bubble. In the Middle Eastern region, it is an ancient type of non-cigarette tobacco smoking that has been generally found. Hookah smoke has more than 4800 numerous substances. Many of them are carcinogens and tumor promoters [5-7]. By reviewing the literature, three flavored Moâssel used in the Egyptian Hookah were analyzed by Head Space GC/MS. They showed great difference in some of them in their composition. Consequently, the harmful effects may be varied from one sample to another [8]. Therefore, the authors provoked to analyze other samples of flavored Moâssel used in in the Egyptian Hookah by Head space GC/MS.

2. Materials and Methods

2.1. Materials

Flavored Egyptian Moâssel specimens *viz.*, Guava (GFM), Mixed Fruits (MixFFM) and Watermelon (WFM) were collected June 2016 from the Egyptian market. The samples were produced in Egypt by Al Dandash Company.

2.2. Method

Shimadzu GC/MS with Head Space system provided by FID (Flame Ionization Detector), connected to the Mass Spectrometer Model: QP2010Ultra. Total GLC chromatograms and mass spectra were recorded in the electron impact ionization mode at 70 eV, using ACQ Mode of scan from 35 to 500 m/z in 0.3 s. The used column was 0.25 mm in internal diameter, 30 m length, packed with Rtx-MS and 0.25 μ m film thickness. The injected volume was 1.0 μ l, using helium as carrier gas at flow rate 40 ml/min. The analyses were carried out at a programmed temperature; the initial temperature was 40 °C (Kept for 2 min), then increased at a rate 30-50 °C to the final temperature 210 °C (kept for 5 min). Injector and detector had the same temperature 230 °C. The total run time was 45 min and split ratio 1:50 [8].

3. Results

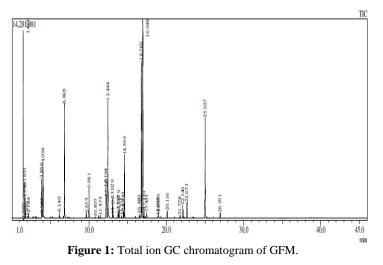
Head Space GC/MS analyses

The qualitative analysis was done by direct comparison of and fragmentation patterns of the identified compounds with archive mass spectra lipid library and quantitation was based on peak area integration [9, 10].

3.1. GFM analysis

GFM exhibited 38 compounds from Head Space GC/MS analysis. The unidentified compounds represented 18.62 % (7 compounds) and identified compounds represented 81.38 % (31

compounds). The major one was fraistone (12.87 %). All identified compounds are oxygenated as shown in (Figure 1) and enumerated in (Table 1).



3.2. MixFFM analysis

Qualitative analysis of MixFFM exhibited 39 compounds using Head Space GC/MS. The unidentified compounds represented 08.02 % (3 compounds) and identified compounds represented 91.98 % (36 compounds). The major one was isoamyl acetate (19.83 %). The identified compounds are classified into two classes *viz.*, 86.67 % oxygenated and 05.31 % hydrocarbons compounds as shown in (**Figure 2**) and enumerated in (**Table 2**).

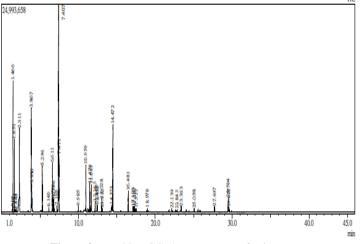


Figure 2: Total ion GC chromatogram of MixFFM.

3.3. WFM analysis

Qualitative analysis WFM exhibited 37 compounds by Head Space GC/MS. The unidentified compounds represented 31.40 % (10 components). The identified compounds represented 68.60 % (27 ingredients). The chief one was 1,2-propanediol (25.23 %). The recognized components are classified into three various classes *viz.*, 60.57 % oxygenated, 04.76 % nitrogenous and 03.27 % hydrocarbons compounds as displayed in (**Figure 3**) and listed in (**Table 3**).

4. Discussion

The current study investigated three different flavored Egyptian Moâssel *viz.*, GFM, MixFFM and WFM by Head Space GC/MS analyses. The samples exhibited very high ratio of oxygenated constituents and traces of (nitrogenous & hydrocarbons) constituents. Therefore, they are powerful flavored samples.

The three samples contained seven common compounds *viz.*, 5,6-epoxy- β -ionone, acetone, acetic acid, 3-methyl-1-butanol, furfural, *Z*-3-hexen-1-ol and benzyl alcohol.

Furthermore, GFM and MixFFM had also seven common compounds *viz.*, 2-methyl-1-butanol, benzaldehyde, *Z*-3-hexenyl acetate, linalool, benzyl acetate, eugenol, *E*-methyl cinnamate. While, GFM and WFM had three common compound; hexanoic acid, anethole and benzyl butanoate. Finally, MixFFM and WFM had one more common compound (isoamyl butyrate). From the aforementioned data, there are relatively differences among the three studied samples, specially between (WFM & MixFFM). But, GFM and MixFFM samples are the most similar samples.

Fraistone (12.87 %) was the major compound in GFM sample. It is colorless liquid with fresh fruity odor. It considered being raw material of manufacturing floral fragrances [11]. Flavor esters compounds have a great commercial importance due to their application chiefly in cosmetic and food industries. Also, they are environmentally benign-solvents and intermediates in pharmaceutical and chemical processes. The most wanted flavor in food industries, are the alkyl esters. Specially, isoamyl acetate due to its strong banana flavor [12]. It is the main compound in MixFFM sample (19.83 %). It is widely used as a flavoring compound in a variety of foodstuffs, such as butterscotch, honey, beverages and artificial coffee [13].

Furthermore, it is also one of the major flavor substance of fermented alcoholic beverages (sake, beer & wines) [13]. On the other hand, it has an irritant and central nervous system depressant effect. Moreover, exposure to concentrations (950 ppm for 30 minutes) caused irritation of the nose & eyes, weakness, headaches, vertigo, palpitations, gastrointestinal disorders, anemia, cutaneous lesions and dermatitis [14]. It may undergo enzymatic hydrolysis to form acetic acid and isoamyl alcohol. Five acetate esters including isoamyl acetate were administered orally to rabbits to study the mechanism of drunkenness. The ester administration reduced the blood pO_2 and pCO_2 [15-17].

The enzymic hydrolysis of isoamyl acetate was studied *in vitro* with preparations of pancreatic and whole homogenates of pig jejunum. Incubation of pancreatin with 500 μ l/l isoamyl acetate and pig jejunum resulted in 100 % hydrolysis. Therefore, it was presumed that those esters hydrolyzed rapidly *in vitro* would also be degraded readily in the animal [15-17].

Finally, the third one WFM showed that 1,2-propanediol (25.23 %) was the main constituents. The 1,2-propanediol (undiluted) was slightly irritating to the eye and producing mild transient conjunctivitis. The eye recovered after the contact removed [18]. Its concentration augmented the hazard of respiratory and immune illnesses in children with hay fever, asthma, allergies and eczema from 50 % to 180 % [19, 20]. The structures of the three main ingredients in the samples are shown in (**Figure 4**).

Ismail *et al.*

Table 1: Identified compounds of GFM from Head Space GC/MS.

| No. | Name | RT* | RRT** | Base peak | Relative Area % | M. Weight | M. Formula |
|-----|--|-------|-------|--------------|--------------------|--------------|--|
| 1 | 5,6-Epoxy- β -ionone | 01.46 | 0.087 | 40 | 06.20 | 208 | $C_{13}H_{20}O_{2}$ |
| 2 | Ethyl lactate | 01.65 | 0.098 | 45 | 01.42 | 118 | $C_5H_{10}O_3$ |
| 3 | Acetone | 01.73 | 0.103 | 43 | 01.20 | 58 | C ₃ H ₆ O |
| 4 | Butanal (syn.: Butyraldehyde) | 01.97 | 0.118 | 43 | 00.31 | 72 | C_4H_8O |
| 5 | Acetic acid | 02.16 | 0.129 | 43 | 00.71 | 60 | $C_2H_4O_2$ |
| 6 | 3-Methyl-1-butanol (Fusel oil) | 03.85 | 0.230 | 55 | 02.25 | 88 | $C_5H_{12}O$ |
| 7 | 2-Methyl-1-butanol | 03.93 | 0.234 | 57 | 00.28 | 88 | $C_5H_{12}O$ |
| 8 | 1,2-Propanediol | 04.04 | 0.241 | 45 | 05.50 | 76 | $C_3H_8O_2$ |
| 9 | Furfural | 06.14 | 0.366 | 96 | 00.43 | 96 | $C_5H_4O_2$ |
| 10 | Z-3-Hexen-1-ol | 06.80 | 0.406 | 41 | 08.60 | 100 | $C_6H_{12}O$ |
| 11 | Ethyl acetoacetate | 09.65 | 0.576 | 43 | 00.38 | 130 | $C_6H_{10}O_3$ |
| 12 | Benzaldehyde | 09.98 | 0.595 | 77 | 02.35 | 106 | C_7H_6O |
| 13 | Hexanoic acid | 10.80 | 0.644 | 60 | 00.17 | 116 | $C_6H_{12}O_2$ |
| 14 | Z-3-Hexenyl acetate | 11.47 | 0.684 | 43 | 00.20 | 142 | $C_8H_{14}O_2$ |
| 15 | 2-Ethyl-1-hexanol | 12.19 | 0.727 | 57 | 02.98 | 130 | $C_8H_{18}O$ |
| 16 | 1,8-Cineole (syn.: Eucalyptol) | 12.24 | 0.730 | 43 | 01.56 | 154 | $C_{10}H_{18}O$ |
| 17 | Benzyl alcohol | 12.44 | 0.742 | 79 | 12.17 | 108 | C_7H_8O |
| 18 | Isoamy butyrate (syn.: Butanoic acid, 3- methylbutyl ester) | 13.03 | 0.777 | 71 | 01.37 | 158 | $C_9H_{18}O_2$ |
| 19 | Amyl butyrate (syn.: Pentyl butanoate) | 13.11 | 0.782 | 71 | 00.25 | 158 | $C_9H_{18}O_2$ |
| 20 | 4-Methyl-2-pentyl, 1,3 dioxolane | 13.88 | 0.828 | 87 | 00.64 | 158 | $C_9H_{18}O_2$ |
| 21 | Linalool | 14.44 | 0.862 | 71 | 00.60 | 154 | $C_{10}H_{18}O$ |
| 22 | Isoamyl isovalerate (syn.: Apple oil) | 14.59 | 0.871 | 70 | 04.50 | 172 | $C_{10}H_{20}O_{2}$ |
| 23 | Benzyl acetate | 16.48 | 0.983 | 108 | 00.33 | 150 | $C_9H_{10}O_2$ |
| 24 | Fraistone (syn.: Ethyl (2,4-Dimethyl- 1,3-dioxolan-2-yl) acetate) | 16.76 | 1.000 | 43 | 12.87 | 188 | C ₉ H ₁₆ O |
| 25 | Z-3-Hexenyl butyrate | 17.11 | 1.021 | 67 | 00.62 | 170 | $C_{10}H_{18}O_2$ |
| 26 | Methyl salicylate | 17.45 | 1.041 | 120 | 00.47 | 152 | C ₈ H ₈ O ₃ |
| 27 | Anethole | 20.11 | 0.200 | | 00.60 | 148 | $C_{10}H_{12}O$ |
| 28 | Benzyl butanoate | 21.78 | 0.300 | 91 | 00.27 | 178 | $C_{11}H_{14}O_2$ |
| 29 | Eugenol | 22.14 | 0.321 | 164 | 01.13 | 164 | $C_{10}H_{12}O_2$ |
| 30 | Z-3-Hexenyl hexoate | 22.67 | 0.353 | 82 | 01.20 | 198 | $C_{12}H_{22}O_{2}$ |
| 31 | E-Methyl cinnamate | 25.02 | 1.493 | 31 | 09.82 | 176 | $C_{11}H_{12}O_2$ |

Oxygenated compounds

*RT: Retention Time. **RRT: Relative Retention Time.

Identified compounds 81.38 %

Ismail *et al.*

Table 2: Identified compounds of MixFFM from Head Space GC/MS.

| No. | Name | RT* | RRT** | Base peak | Relative Area % | M. Weight | M. Formula |
|-------|--|-------|-------|--------------|--------------------|--------------|---------------------------------|
| 1 | 5,6-Epoxy-β-ionone | 1.46 | 0.197 | 39 | 6.11 | 208 | $C_{13}H_{20}O_2$ |
| 2 | Ethanol | 1.65 | 0.222 | 45 | 2.70 | 46 | C_2H_6O |
| 3 | Acetone | 1.73 | 0.223 | 43 | 0.37 | 58 | C ₃ H ₆ O |
| 4 | Acetic acid | 1.85 | 0.250 | 43 | 0.41 | 60 | $C_2H_4O_2$ |
| 5 | Dihydrolinalool | 2.31 | 0.312 | 43 | 4.59 | 88 | $C_4H_8O_2$ |
| 6 | 3-Methyl-1-butanol (Fusel oil) | 3.86 | 0.521 | 55 | 8.15 | 88 | $C_5H_{12}O$ |
| 7 | 2-Methyl-1-butanol | 3.93 | 0.531 | 41 | 2.44 | 88 | $C_5H_{12}O$ |
| 8 | Ethyl butanoate (syn.: Ethyl butyrate) | 5.29 | 0.714 | 71 | 3.56 | 116 | $C_6H_{12}O_2$ |
| 9 | Furfural | 6.14 | 0.829 | 96 | 0.42 | 96 | $C_5H_4O_2$ |
| 10 | Ethyl-2-methyl butanoate | 6.61 | 0.893 | 57 | 3.99 | 130 | $C_7H_{14}O_2$ |
| 11 | Ethyl-3-methyl butanoate | 6.70 | 0.905 | 88 | 1.07 | 130 | $C_7H_{14}O_2$ |
| 12 | Z-3-Hexen-1-ol | 6.79 | 0.917 | 41 | 1.44 | 100 | $C_6H_{12}O$ |
| 13 | <i>n</i> -Hexyl formate | 7.19 | 0.971 | 56 | 0.72 | 130 | $C_7H_{14}O_2$ |
| 14 | Isoamyl acetate | 7.40 | 1.000 | 43 | 19.83 | 130 | $C_7H_{14}O_2$ |
| 15 | 2-Methyl butyl acetate | 7.47 | 1.009 | 43 | 4.54 | 130 | $C_7H_{14}O_2$ |
| 16 | Benzaldehyde | 9.98 | 1.348 | 77 | 0.64 | 106 | C ₇ H ₆ O |
| 17 | Myrcene | 10.95 | 1.479 | 41 | 4.31 | 136 | $C_{10}H_{16}$ |
| 18 | Z-3-Hexenyl acetate | 11.47 | 1.550 | 43 | 2.70 | 142 | $C_8H_{14}O_2$ |
| 19 | <i>n</i> -Hexyl acetate | 11.68 | 1.578 | 43 | 2.62 | 144 | $C_8H_{16}O_2$ |
| 20 | Limonene | 12.16 | 1.643 | 68 | 1.00 | 136 | $C_{10}H_{16}$ |
| 21 | Benzyl alcohol | 12.40 | 1.675 | 79 | 0.83 | 108 | C ₇ H ₈ O |
| 22 | Isoamyl butyrate | 13.02 | 1.759 | 71 | 1.61 | 158 | $C_9H_{18}O_2$ |
| 23 | Glycerol monoacetate | 14.32 | 1.935 | 43 | 0.96 | 134 | $C_5H_{10}O_4$ |
| 24 | Linalool | 14.47 | 1.955 | 71 | 9.79 | 154 | $C_{10}H_{18}O$ |
| 25 | Benzyl acetate | 16.48 | 2.227 | 108 | 0.20 | 150 | $C_9H_{10}O_2$ |
| 26 | Z-3-Hexenyl butyrate | 17.10 | 2.310 | 67 | 0.55 | 170 | $C_{10}H_{18}O_2$ |
| 27 | Hexyl butanoate | 17.26 | 2.332 | 43 | 0.61 | 172 | $C_{10}H_{20}O_2$ |
| 28 | Ethyl maltol | 17.52 | 2.367 | 140 | 0.42 | 140 | $C_7H_8O_3$ |
| 29 | Isoamyl caproate | 18.98 | 2.564 | 70 | 0.42 | 186 | $C_{11}H_{22}O_2$ |
| 30 | Eugenol | 22.14 | 2.991 | 164 | 0.41 | 164 | $C_{10}H_{12}O_2$ |
| 31 | E-Methyl cinnamate | 22.84 | 3.086 | 131 | 0.33 | 162 | $C_{10}H_{10}O_2$ |
| 32 | Diphenylether | 23.36 | 3.156 | 170 | 0.87 | 170 | $C_{12}H_{10}O$ |
| 33 | γ-Decalactone | 25.06 | 3.386 | 85 | 0.42 | 170 | $C_{10}H_{18}O_2$ |
| 34 | γ-Undecalactone | 27.69 | 3.741 | 85 | 0.74 | 184 | $C_{11}H_{20}O_2$ |
| 35 | Jasmal | 29.30 | 3.986 | 129 | 1.67 | 214 | $C_{12}H_{22}O_3$ |
| 36 | Dihydro methyl jasmonate | 29.60 | 4.000 | 83 | 0.54 | 226 | $C_{13}H_{22}O_{3}$ |
| Unide | ntified compounds 08.02% | | | | | | |

*RT: Retention Time. **RRT: Relative Retention Time.

Ismail *et al.*

Table 3: Identified compounds of WFM from Head Space GC/MS.

| No. | Name | RT* | RRT** | Base peak | Relative Area % | M. Weight | M. Formula |
|-----|---|-------------|-------|--------------|--------------------|--------------|---------------------------------|
| 1 | 5,6-Epoxy-β-ionone | 1.46 | 0.350 | 39 | 9.50 | 208 | $C_{13}H_{20}O_2$ |
| 2 | Nitrosamine | 1.67 | 0.401 | 45 | 4.76 | 46 | H_2N_2O |
| 3 | Acetone | 1.73 | 0.415 | 43 | 0.56 | 58 | C ₃ H ₆ O |
| 4 | Acetic acid | 2.22 | 0.533 | 43 | 0.88 | 60 | $C_2H_4O_2$ |
| 5 | Hydroxyacetone (syn.: 1-Hydroxy-2- propanone) | 2.83 | 0.680 | 43 | 0.43 | 74 | $C_3H_6O_2$ |
| 6 | 3-Methyl-1-butanol (Fusel oil) | 3.85 | 0.925 | 55 | 3.10 | 88 | $C_5H_{12}O$ |
| 7 | 1,2-Propanediol | 4.16 | 1.00 | 45 | 25.23 | 76 | $C_3H_8O_2$ |
| 8 | Butanoic acid | 5.16 | 1.240 | 60 | 0.66 | 88 | $C_4H_8O_2$ |
| 9 | Ethyl butanoate (syn.: Ethyl butyrate) | 5.30 | 1.274 | 71 | 0.87 | 116 | $C_6H_{12}O_2$ |
| 10 | Furfural | | | 96 | 0.58 | 96 | $C_5H_4O_2$ |
| 11 | Z-3-Hexen-1-ol | 6.79 | 1.632 | 41 | 1.23 | 100 | $C_6H_{12}O$ |
| 12 | α-Phellandrene | 8.90 | 2.139 | 93 | 0.46 | 136 | $C_{10}H_{16}$ |
| 13 | <i>α</i> -Pinene | 9.11 | 2.189 | 93 | 0.73 | 136 | $C_{10}H_{16}$ |
| 14 | β -Phellandrene | 10.39 | 2.497 | 93 | 0.50 | 136 | $C_{10}H_{16}$ |
| 15 | β-Pinene | 10.49 | 2.521 | 93 | 0.51 | 136 | $C_{10}H_{16}$ |
| 16 | Hexanoic acid | 10.70 | 2.572 | 60 | 0.44 | 116 | $C_6H_{12}O_2$ |
| 17 | α-Terpinene | 11.77 | 2.829 | 121 | 0.49 | 136 | $C_{10}H_{16}$ |
| 18 | Sylvestrene | 12.17 | 2.925 | 93 | 0.58 | 136 | $C_{10}H_{16}$ |
| 19 | Benzyl alcohol | 12.41 | 2983 | 79 | 1.79 | 108 | C_7H_8O |
| 20 | 2,6-Dimethylhept-5-en-1-al | 12.97 | 3.117 | 82 | 2.59 | 140 | $C_9H_{16}O$ |
| 21 | Isoamyl butyrate | 13.12 | 3.153 | 71 | 3.46 | 158 | $C_9H_{18}O_2$ |
| 22 | Acetophenone | 13.40 | 3.221 | 104 | 1.14 | 120 | C ₈ H ₈ O |
| 23 | Menthol | 16.76 | 4.028 | 71 | 1.35 | 156 | $C_{10}H_{20}O$ |
| 24 | Anethole | 20.11 | 4.834 | 148 | 1.54 | 148 | $C_{10}H_{12}O$ |
| 25 | Benzyl butanoate | 21.78 | 5.048 | 91 | 1.26 | 178 | $C_{11}H_{14}O_2$ |
| 26 | Vanillin | 23.31 | 5.603 | 151 | 0.50 | 152 | $C_8H_8O_3$ |
| 27 | Cinnamyl isovalerate | 30.27 | 7.276 | 85 | 3.46 | 218 | $C_{14}H_{18}O_2$ |
| | - | xygenated c | - | 0.57 % | | | |
| | Nitrogenous compounds 04.76 % Hydrocarbons compounds 03.27 % | | | | | | |

*RT: Retention Time. **RRT: Relative Retention Time.

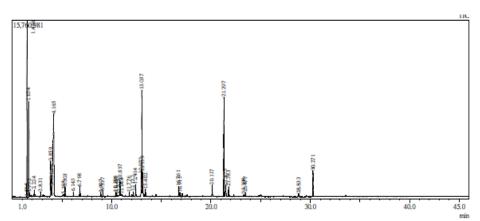


Figure 3: Total ion GC chromatogram of WFM.

81

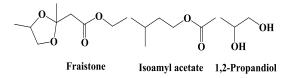


Figure 4: The major constituent for each sample.

5. Conclusion

The comparative study of Guava, Mixed Fruits and Watermelon Moâssel from Al Dandash Company (Egypt), exhibited relatively differences in the chemical constituents. Therefore, this study may introduce a toxicological effects prediction for these specimens.

6. Conflict of Interest

We declare that no conflict of interest.

References

[1] Doll R, Peto R, Boreham J, Sutherland I. Mortality in relation to smoking: 50 years observations on male British doctors. *British Medical Journal*. 2004;328:1519-33.

[2] Pirie K, Peto R, Reeves GK, Green J, Beral V. The 21st century hazards of smoking and benefits of stopping: a prospective study of one million women in the UK. *Lancet.* 2013;381:133-41.

[3] www.rcplond on.ac.uk/sites/default/files/documents/passive-smoking-andchildren-pdf (Accessed 25 February 2016) Royal College of Physicians, Passive smoking and children. A report by Tobacco advisory group of the Royal College of Physicians. London (Retrieved 05.07.2017).

[4] https://cdn.shopify.com/s/files/1/0924/4392/files/going-smok-free

pdf.2801907981964551469 (Accessed 25 February 2016) Royal Collage of Physicians. Going smoke-free:The medical case for clear air in home, at work and in public places. A report on passive smoking by the Tobacco Advisory group of-the Royal Collage Physicians. London: RCP, 2005.

[5] Dar-Odeh NS, Abu-Hammad OA. Narghile smoking and its adverse health consequences: a literature review. *British Dental Journal*. 2009;206(11):571-3.

[6] Sajid KM, Chaouach K, Mahmood R. Hookah Smoking and Cancer: Carcino-embryonic antigen (CEA) levels in exclusive/ever hookah smokers. *Harm Reduction Journal*. 2008;5(19):1-14.

[7] Hoffmann D, Hoffmann I, El-Bayoumy K. The Less Harmful Cigarette: a controversial issue. a tribute to Ernst L. Wynder. *Chemical Research in Toxicology*. 2001;14(7):767-90.

[8] Ismail MM, Hamed ANE, Fouad MA, Kamel MS. Comparative Head Space GC/MS Studies of Different Flavored Moâssel in the Egyptian Market (I). *International Journal of Pharmacognosy and Phytochemical Research*. 2018;10(3):116-22.

[9] National Institute of Standards and Technology (NIST); http://webbook.nist.gov/ (Retrieved 15.09.2016).

[10] Adams RP. Identification of essential oil components by Gas Chromatography/Mass Spectrometry. Edn. 4th, Illinois, USA, Allured books, 1989.

[11] Surburg H, Panten J. Common Fragrance and Flavor Material Book 5th Ed., Individual Fragrance and Flavor Material, 2006,162.

[12] Torres S, Baigorí MD, Pandey A, Castro GR. Production and Purification of a solvent-resistant esterase from *Bacillus licheniformis* S-86. *Applied Biochemistry and Biotechnology Journal*. 2008;151:221-32.

[13] Welsh FW, Williams RE, Dawson KH. Lipase mediated synthesis of low molecular weight flavor esters. *Journal of Food Science*. 1990;55;1679-82.

[14] Opdyke DLJ. Monographs on fragrance raw materials: Isoamyl acetate. *Food and Cosmetics Toxicology Journal*. 1975;13(5):545-54.

[15] Sandmeyer EE, Kirwin CJ. Esters In: Clayton, G.D. & Clayton, F.E., Eds., Patty's Industrial Hygiene and Toxicology, 3rd Edn., John Wiley and Sons, New York, 1981;II(A):2259-412.

[16] Grundschober F. Toxicological assessment of flavoring esters. *Toxicology*. 1978;8(3):387-90.

[17] Tambo S. Toxicity hazard of paint thinners with particular emphasis on the metabolism and toxicity of acetate esters. *Nichidai Igaku Zasshi*. 1973;32(3):349-60.

[18] Robertson OH, Loosli CG, Puck TT, Wise H, Lemon HM, Lester W. Tests for the chronic toxicity of propylexe glycol and triethylene glycol on monkeys and rats by vapor inhalation and oral administration. *The Journal of Pharmacology and Experimental Therapeautics*. 1947;91(1):52-76.

[19] Choi H, Schmidbauer N, Sundell J, Hasseigren M, Spengler J, Bornehag CG. Common Household Chemicals and the Allergy Risks in Pre-School Age Children. *PloS ONE*. 2010;5(10):e13423.

[20] http://journals.plos.org/plosone/article?id=10.1371/journal.pone.0013423#p one.0013423-Chalubinski1, Chemical Compounds Emitted from Common Household Pants and Cleaners Increase Risks of Asthma and Allergies in Children, 2010, (Retrieved 22.07.2017).